

# **Oxazepines and Thiazepines, XXVIII [1] Circular Dichroism of Optically Active 2,3- Dihydro-2-methyl-1,5-benzoxazepin-4(5*H*)-ones\*\*\***

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**Summary.** Chiroptical properties of the title compounds have been studied. The influence of the substitution pattern of the aromatic moiety and the consequence of the amide → thioamide conversion are discussed as well.

**Keywords.** Chiroptical properties; Circular dichroism; Cotton effects.

**Oxazepine und Thiazepine, 28. Mitt.:**

**Circulardichroismus optisch aktiver 2,3-Dihydro-2-methyl-1,5-benzoxazepin-4(5*H*)-one**

**Zusammenfassung.** Die chiroptischen Eigenschaften der Titelverbindungen wurden untersucht. Der Einfluß des Substitutionsmusters am aromatischen Ring sowie die Folgen der Amid → Thioamid-Umsetzung werden diskutiert.

## **Introduction**

The first optically active benzoxazepine was prepared by the reaction of a sugar derivative with 2-aminophenol [2]. Later, optically active 1,5-benzoxazepine carboxylic acids possessing angiotensin converting enzyme inhibitor activity were synthesized [3]. Recently Schultz et al. [4, 5] reported the synthesis of optically active 1,4-benzoxazepine derivatives. In the course of our studies on benzoxazepines we prepared a large series of optically active 2,3-dihydro-2-methyl-1,5-benzoxazepin-4(5*H*)-ones [6]. In our present paper the circular dichroism of these benzoxazepines is reported. The circular dichroism of analogous benzodiazepines [7–10] and benzothiazepines [11–13] was investigated in detail but, to our knowledge, no chiroptical data of benzoxazepines have hitherto been described.

\*\* Deceased on January 14, 1992

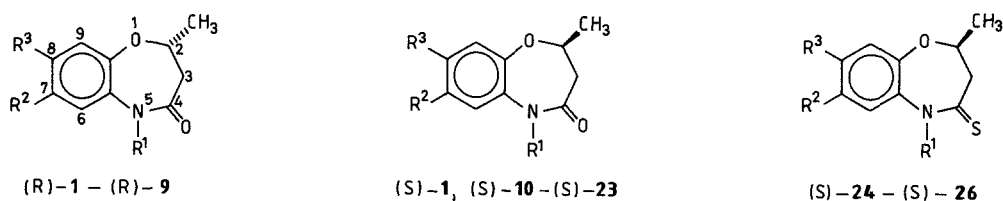
\*\*\* Dedicated to Prof. Dr. W. Wiegrebe on the occasion of his 60th birthday

## Results and Discussion

Parent compounds of the compounds under investigation are the enantiomers (*R*)-**1** and (*S*)-**1** of 2,3-dihydro-2-methyl-1,5-benzoxazepin-4(*5H*)-one, the chromophore of which is a benzene ring perturbed by an ether oxygen and a lactam nitrogen. In their UV-spectra three distinct bands with a shoulder (281 nm) are observed (Table 1). In their CD-spectra two maxima can be assigned to each UV-band. In the case of the (*R*)-**1** a weak negative (289 nm) and a weak positive (282 nm) Cotton effect belonging to the  $\alpha$ -band can be detected. The rotatory strength of the

**Table 1.** UV spectroscopic data

| Compound   | $R^1$   | $R^2$                             | $R^3$                                | $\lambda$ [nm] ( $\epsilon$ )                    |
|--|---|-----------------------------------|--------------------------------------|--|
| ( <i>R</i> )- <b>1</b> }<br>( <i>S</i> )- <b>1</b> } | H   | H                                 | H                                    | 209 (34890), 242 (8540), 275 (1960), 281 (1850)  |
| ( <i>R</i> )- <b>2</b>                               | H   | C <sub>2</sub> H <sub>5</sub>     | H                                    | 212 (35290), 242 (9380), 282 (2420), 286 (2430)  |
| ( <i>R</i> )- <b>3</b>                               | H   | CH <sub>3</sub> O                 | H                                    | 215 (46700), 240 (9370), 292 (5450)              |
| ( <i>R</i> )- <b>4</b>                               | H   | OH                                | H                                    | 213 (32090), 240 (7200), 292 (4200)              |
| ( <i>R</i> )- <b>5</b>                               | H   | CH <sub>3</sub> COO               | H                                    | 212 (30690), 241 (8850), 283 (2690)              |
| ( <i>R</i> )- <b>6</b>                               | H   | COOH                              | H                                    | 230 (32310), 258 (8220), 298 (2310)              |
| ( <i>R</i> )- <b>7</b>                               | H   | CH <sub>3</sub> CO                | H                                    | 238 (31100), 266 (9410), 304 (2830)              |
| ( <i>R</i> )- <b>8</b>                               | CH <sub>3</sub> CO                            | H                                 | H                                    | 201 (34280), 225 (9940), 268 (1170), 276 (890)   |
| ( <i>R</i> )- <b>9</b>                               | CH <sub>3</sub>                               | CH <sub>3</sub> O                 | H                                    | 217 (27850), 245 (7400), 291 (4190)              |
| ( <i>S</i> )- <b>10</b>                              | H   | CH <sub>3</sub>                   | H                                    | 212 (30570), 242 (8000), 281 (2280), 286 (2310)  |
| ( <i>S</i> )- <b>11</b>                              | H   | (CH <sub>3</sub> ) <sub>3</sub> C | H                                    | 212 (36810), 243 (8850), 279 (2590), 285 (2590)  |
| ( <i>S</i> )- <b>12</b>                              | H   | NH <sub>2</sub>                   | H                                    | 224 (27660), 250 (6770), 310 (3110)              |
| ( <i>S</i> )- <b>13</b>                              | H   | Cl                                | H                                    | 216 (38030), 245 (8470), 285 (2540), 290 2600    |
| ( <i>S</i> )- <b>14</b>                              | H   | NO <sub>2</sub>                   | H                                    | 232 (12350), 252 (15980), 295 (5360), 334 (3920) |
| ( <i>S</i> )- <b>15</b>                              | H   | H                                 | NH <sub>2</sub>                      | 208 (23400), 260 (12710), 302 (2850)             |
| ( <i>S</i> )- <b>16</b>                              | H   | H                                 | (CH <sub>3</sub> ) <sub>2</sub> CHNH | 206 (25120), 269 (19020), 308 (3690)             |
| ( <i>S</i> )- <b>17</b>                              | H   | H                                 | CH <sub>3</sub> CONH                 | 213 (28230), 264 (20230), 294 (5000)             |
| ( <i>S</i> )- <b>18</b>                              | H   | H                                 | Cl                                   | 212 (27790), 249 (11980), 285 (2340)             |
| ( <i>S</i> )- <b>19</b>                              | H   | H                                 | Br                                   | 213 (30600), 251 (13410), 283 (2690)             |
| ( <i>S</i> )- <b>20</b>                              | CH <sub>3</sub>                               | H                                 | H                                    | 211 (28410), 244 (8990), 276 (2230)              |
| ( <i>S</i> )- <b>21</b>                              | C <sub>2</sub> H <sub>5</sub>                 | H                                 | H                                    | 211 (24630), 244 (9220), 276 (1910)              |
| ( <i>S</i> )- <b>22</b>                              | C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub> | H                                 | H                                    | 194 (38800), 209 (28750), 244 (8450), 276 (1830) |
| ( <i>S</i> )- <b>23</b>                              | CH <sub>3</sub>                               | H                                 | NO <sub>2</sub>                      | 203 (19950), 233 (7090), 326 (9570)              |
| ( <i>S</i> )- <b>24</b>                              | H   | H                                 | H                                    | 204 (17800), 308 (20660)                         |
| ( <i>S</i> )- <b>25</b>                              | H   | (CH <sub>3</sub> ) <sub>3</sub> C | H                                    | 208 (28570), 310 (24910)                         |
| ( <i>S</i> )- <b>26</b>                              | CH <sub>3</sub>                               | H                                 | H                                    | 201 (22310), 298 (23190)                         |



CD-maxima in the regions of the *p*-band (at 257 and 237 nm) and the  $\beta$ -band (at 216 and 200 nm) is much higher. The same type was observed for the enantiomorphous CD-spectrum of the (*S*)-1 enantiomer (see Fig. 1 and Table 2). The influence of substituents at C-7 and C-8 and N-alkylation, respectively, have been studied as well.

Introduction of a methyl [(*S*)-10], ethyl [(*R*)-2] or *tert*-butyl [(*S*)-11] group into position 7 is almost without influence both on the UV- and the CD-spectra. The presence of a chlorine atom [(*S*)-13] or an acetoxy group [(*R*)-5] resulted only in a little shift of the CD-maximum to the longer wavelengths. A methoxy [(*R*)-3] and a hydroxy [(*R*)-4] group in position 7 are stronger perturbers but the basic character of the CD-spectra is unchanged. The rotatory strength in the  $\beta$ -region was considerably enhanced by a C-7 carboxylic group [(*R*)-6]. The presence of an acetyl group in this position [(*R*)-7] resulted in an additional  $n \rightarrow \pi^*$  transition giving rise to a new CD-band at 324 nm (Table 2). An amino group at C-7 [(*S*)-12] resulted also in considerable changes in all three regions of the CD-spectrum. The influence of a C-7 nitro group [(*S*)-14] is reflected in the complete change of the character of the chromophore and chiroptical properties.

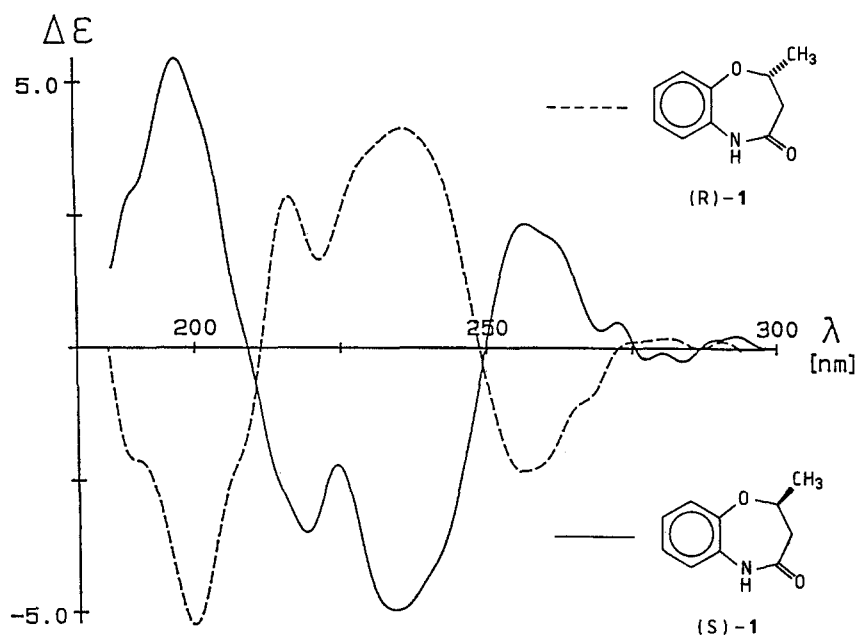


Fig. 1. Circular dichroism spectra of compounds (*R*)-1 and (*S*)-1

**Table 2.** Circular dichroism data

| Compound | $\lambda$ [nm]( $\Delta\epsilon$ )   |
|----------|--|
| (R)-1    | 200 (-5.22), 216 (+2.84), 237 (+4.45), 257 (-2.43), 282 (+0.07), 289 (-0.06)                 |
| (R)-2    | 204 (-4.02), 220 (+2.93), 234 (+3.40), 239 (+3.50), 258 (-1.70), 286 (+0.21),<br>295 (-0.05) |
| (R)-3    | 238 (+5.10), 257 (-1.95), 290 (+0.35), 297 (+0.37) 307 (-0.09)                               |
| (R)-4    | 239 (+3.79), 258 (-1.37), 297 (+0.35)  |
| (R)-5    | 199 (-3.18), 215 (+3.16), 223 (+2.62), 236 (+3.35), 257 (-2.08), 287 (-0.17),<br>295 (-0.06) |
| (R)-6    | 237 (+7.43), 269 (-2.62), 294 (+0.05)  |
| (R)-7    | 240 (+4.91), 270 (-1.89), 324 (+0.24)  |
| (R)-8    | 197 (-7.43), 230 (+2.32), 259 (-0.34), 278 (+0.15)   |
| (R)-9    | 241 (+3.19), 298 (+0.79)   |
| (S)-1    | 197 (+5.44), 219 (-3.47), 235 (-4.24), 257 (+2.15), 282 (-0.11), 291 (+0.08)                 |
| (S)-10   | 199 (+3.38), 203 (+3.25), 217 (-2.53), 235 (-3.63), 258 (+1.66), 280 (-0.18),<br>287 (+0.25) |
| (S)-11   | 211 (-8.05), 235 (-4.78), 257 (+2.58), 280 (-0.08), 285 (-0.18), 292 (+0.06)                 |
| (S)-12   | 229 (-4.68), 265 (+0.48), 314 (-0.47)  |
| (S)-13   | 204 (+3.55), 221 (-3.87), 240 (-3.30), 259 (+2.47), 284 (-0.18), 290 (-0.23)                 |
| (S)-14   | 216 (+1.99), 232 (-1.52), 277 (+0.22), 318 (-0.42)   |
| (S)-15   | 204 (+6.92), 254 (-2.98), 307 (-0.17)  |
| (S)-16   | 209 (+6.33), 234 (-0.93), 265 (-2.61), 308 (-0.15)   |
| (S)-17   | 247 (-2.18), 276 (+0.93)   |
| (S)-18   | 218 (-2.11), 240 (-3.49), 263 (+1.73), 289 (-0.07)   |
| (S)-19   | 223 (-2.55), 240 (-3.54), 264 (+1.80), 291 (-0.07), 301 (+0.04)                              |
| (S)-20   | 196 (+3.79), 202 (+4.53), 242 (-4.27), 276 (-0.59)   |
| (S)-21   | 202 (+3.61), 239 (-3.45), 277 (-0.53)  |
| (S)-22   | 201 (+2.77), 215 (-2.55), 238 (-2.89), 276 (-0.42), 281 (-0.39)                              |
| (S)-23   | 207 (+2.83), 301 (-0.73)   |
| (S)-24   | 188 (+7.75), 208 (-4.74), 233 (+7.82), 306 (-8.03), 383 (+1.08)                              |
| (S)-25   | 205 (-11.74), 233 (+7.51), 260 (-1.15), 308 (-9.26), 342 (-1.07), 382 (+1.54)                |
| (S)-26   | 189 (+10.80), 218 (-3.65), 234 (+6.91), 296 (-7.07), 367 (-1.94)                             |

The presence of a substituent at C-8 resulted in a considerable intensity increase of the *p*-band of the UV-spectra of each compound (Table 1). A chlorine [(S)-18] or a bromine [(S)-19] atom at position 8 is almost without influence on the chiroptical properties. However, the introduction of an amino [(S)-15] or substituted amino [(S)-16] and [(S)-17] group into this position resulted in considerable changes of the whole spectrum.

N-Alkylation causes little change in the UV-spectra since the chromophore remains unaltered, whereas in the CD-spectra of compounds (S)-20, (S)-21, and (S)-22 the rotatory strength of the *p*-band is considerably enhanced. This may be a consequence of the alteration in the conformational equilibrium as a consequence of the N-alkylation [14].

As a result of the amide  $\rightarrow$  thioamide conversion, the 'three-band character' of the UV-spectra disappeared and entirely new spectra were obtained with two intense

bands: One between 300 and 310 nm may belong to the  $\pi \rightarrow \pi^*$  transition of the C=S group and the other (at approx. 200 nm) to the electron transitions of the aromatic  $\pi$  orbitals. In their CD-spectra at least five Cotton effects were found. In the case of compounds (*S*)-**24** and (*S*)-**25** an intense positive CD-band appears at approx. 380 nm which may belong to the  $n \rightarrow \pi^*$  transition of thioamide moiety. The second intense negative CD-band around 300 nm corresponds to the UV-maximum of substances (*S*)-**24**, (*S*)-**25**, and (*S*)-**26** and may be a consequence of an electronically allowed  $\pi \rightarrow \pi^*$  transition of the thioamide chromophore. Three other Cotton-effects are found in the shorter wavelength region (Table 2) but their unambiguous assignment is impossible at present. The presence of a *tert*-butyl group at C-7 [(*S*)-**25**] and N-methylation [(*S*)-**26**] are almost without influence on the CD-spectra.

### Experimental Part

The compounds investigated were synthesized as described earlier [6]. UV-spectra were measured with a Philips PU 8740 apparatus in CH<sub>3</sub>CN solution at room temperature. Circular dichroism (CD) spectra were recorded with a Jasco 600 instrument in CH<sub>3</sub>CN solution (concentrations were approx. 0.5 mmol/l both for UV- and CD-measurements) at room temperature.

### Acknowledgements

The present study was sponsored by the Deutsche Forschungsgemeinschaft (DFG) and by the Hungarian Academy of Sciences (Grant No. OTKA-1696) for which our gratitude is expressed. J. O. is grateful to the DFG for a fellowship.

### References

- [1] Part XXVII: Lévai A., Bálint Z. Arch. Pharm. (Weinheim), in press
- [2] Tronchet J. M. J., Gentile B. (1980) Helv. Chim. Acta **63**: 1779
- [3] Itoh K., Kori M., Inada Y., Nishikawa K., Kawamatsu Y., Sugihara H. (1986) Chem. Pharm. Bull. **34**: 1128, 2078, 3747
- [4] Schultz A. G., Pinto D. J. P., Welch M. (1988) J. Org. Chem. **53**: 1372
- [5] Schultz A. G., Macielag M., Sundararaman P., Teveras A. G., Welch M. (1988) J. Am. Chem. Soc. **110**: 7828
- [6] Lévai A., Ott J., Snatzke G. (1992) Monatsh. Chem. **123**: 919
- [7] Alebič-Kolbah T., Kajfez F., Rendič S., Sunjič V., Konowal A., Snatzke G. (1979) Biochem. Pharmacol. **28**: 2457
- [8] Konowal A., Snatzke G., Alebič-Kolbah T., Kajfez F., Rendič S., Sunjič V. (1979) Biochem. Pharmacol. **28**: 3109
- [9] Snatzke G., Konowal A., Sabljic A., Blazevič N., Sunjič V. (1982) Croat. Chem. Acta **55**: 435
- [10] Kajtár M., Kajtár J., Röhricht J., Ángyán J. G. (1989) Croat. Chem. Acta **62**: 245
- [11] Kojič-Prodič B., Ruzič-Toros Z., Sunjič V., Decorte E., Moimas F. (1984) Helv. Chim. Acta **67**: 916
- [12] Ciechanowicz-Rutkowska M., Grochowski J., Lévai A., Puzicha G., Serda P., Snatzke G. (1989) Monatsh. Chem. **120**: 981
- [13] Puzicha G., Lévai A., Snatzke G. (1990) Monatsh. Chem. **121**: 293
- [14] Ott J., Hiegemann M., Duddeck H. (1991) Magn. Reson. Chem. **29**: 244

Received February 26, 1992. Accepted March 26, 1992